

CHEM 3331 (Richardson) Midterm Exam 3 – Apr. 16, 2019

Your Name: _____

Student ID: _____

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|-----------------------------|-----------------------------|
| Recitation (check one) | O 149 (Thu 5:00 w/ Will) |
| O 130 (Wed 8:00 w/ Olivia) | O 235 (Wed 1:00 w/ Lauren) |
| O 134 (Wed 12:00 w/ Olivia) | O 237 (Wed 3:00 w/ Lauren) |
| O 136 (Wed 2:00 w/ Lacey) | O 239 (Wed 5:00 w/ Zepeng) |
| O 138 (Wed 4:00 w/ Lacey) | O 240 (Thu 8:00 w/ Zhenhao) |
| O 141 (Thu 9:00 w/ Chance) | O 242 (Thu 10:00 w/ Lauren) |
| O 143 (Thu 11:00 w/ Chance) | O 244 (Thu 12:00 w/ Lauren) |
| O 145 (Thu 1:00 w/ Lacey) | O 246 (Thu 2:00 w/ Brianna) |
| O 147 (Thu 3:00 w/ Will) | O 248 (Thu 4:00 w/ Brianna) |

Question	Score	Out of
1	30	
2	15	
3	15	
4	30	
5	10	
6	10 e.c.	
Total		100

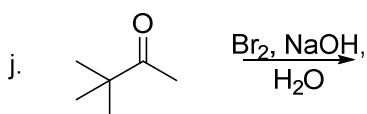
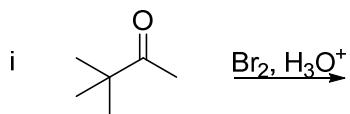
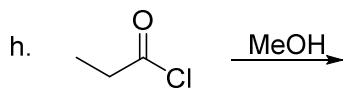
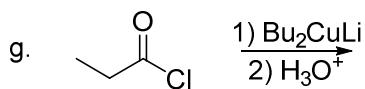
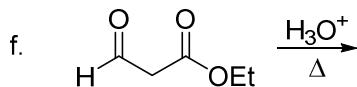
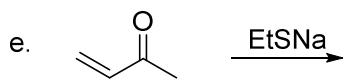
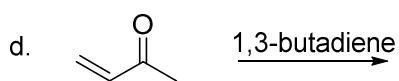
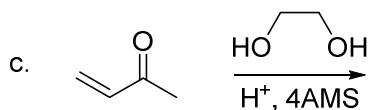
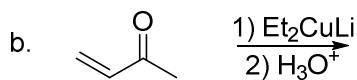
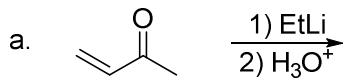
This is a closed-book exam. The use of notes, calculators, or cell phones will not be allowed during the exam. You may use models sets brought in a clear bag. Use the backs of the pages for scratch work. If your final answer is not clearly specified, you will lose points. For mechanisms, show all intermediates including correct formal charges, but do not show transition states.

Periodic Table of the Elements																	
1 IA 1A	2 IIA 2A	3 Li Lithium 6.941	4 Be Beryllium 9.012	5 VB 5B	6 VIB 6B	7 VIIIB 7B	8	9 VII 8	10	11 IB 1B	12 IIB 2B	13 IIIA 3A	14 IVA 4A	15 VA 5A	16 VIA 6A	17 VIIA 7A	18 VIIIA 8A
1 H Hydrogen 1.008												5 B Boron 10.811	6 C Carbon 12.011	7 N Nitrogen 14.007	8 O Oxygen 15.999	9 F Fluorine 18.998	2 He Helium 4.003
11 Na Sodium 22.990	12 Mg Magnesium 24.305	3 IIIIB 3B	4 IVB 4B	5 VB 5B	6 VIB 6B	7 VIIIB 7B				10 IB 1B	11 IIB 2B	13 Al Aluminum 26.982	14 Si Silicon 28.086	15 P Phosphorus 30.974	16 S Sulfur 32.066	17 Cl Chlorine 35.453	18 Ar Argon 39.948
19 K Potassium 39.098	20 Ca Calcium 40.078	21 Sc Scandium 44.961	22 Ti Titanium 47.88	23 V Vanadium 50.942	24 Cr Chromium 51.996	25 Mn Manganese 54.938	26 Fe Iron 55.845	27 Co Cobalt 58.933	28 Ni Nickel 58.693	29 Cu Copper 63.546	30 Zn Zinc 65.38	31 Ga Gallium 69.723	32 Ge Germanium 72.031	33 As Arsenic 74.922	34 Se Selenium 78.971	35 Br Bromine 79.904	36 Kr Krypton 84.78
37 Rb Rubidium 85.468	38 Sr Strontium 87.62	39 Y Yttrium 88.906	40 Zr Zirconium 91.224	41 Nb Niobium 92.906	42 Mo Molybdenum 95.95	43 Tc Technetium 98.907	44 Ru Ruthenium 101.07	45 Rh Rhodium 102.906	46 Pd Palladium 106.42	47 Ag Silver 107.868	48 Cd Cadmium 112.414	49 In Indium 114.818	50 Sn Tin 118.711	51 Sb Antimony 121.760	52 Te Tellurium 127.6	53 I Iodine 126.904	54 Xe Xenon 131.294
55 Cs Cesium 132.905	56 Ba Barium 137.328	57-71	72 Hf Hafnium 178.49	73 Ta Tantalum 180.948	74 W Tungsten 183.85	75 Re Rhenium 186.207	76 Os Osmium 190.23	77 Ir Iridium 192.22	78 Pt Platinum 195.08	79 Au Gold 196.957	80 Hg Mercury 200.59	81 Tl Thallium 204.383	82 Pb Lead 207.2	83 Bi Bismuth 208.980	84 Po Polonium [208.982]	85 At Astatine 209.987	86 Rn Radium 222.018
87 Fr Francium 223.020	88 Ra Radium 226.025	89-103	104 Rf Rutherfordium [261]	105 Db Dubnium [262]	106 Sg Seaborgium [263]	107 Bh Bohrium [264]	108 Hs Hassium [265]	109 Mt Meitnerium [268]	110 Ds Darmstadtium [270]	111 Rg Roentgenium [285]	112 Cn Copernicium [285]	113 Nh Nihonium [289]	114 Fl Florium [289]	115 Mc Moscovium [289]	116 Lv Livermorium [293]	117 Ts Tennessine [293]	118 Og Oganesson [294]
Lanthanide Series		57 La Lanthanum 138.905	58 Ce Cerium 140.116	59 Pr Praseodymium 140.908	60 Nd Neodymium 144.243	61 Pm Promethium 144.913	62 Sm Samarium 150.36	63 Eu Europium 151.964	64 Gd Gadolinium 157.25	65 Tb Terbium 158.925	66 Dy Dysprosium 162.500	67 Ho Holmium 164.930	68 Er Erbium 167.259	69 Tm Thulium 168.934	70 Yb Ytterbium 173.055	71 Lu Lutetium 174.967	
Actinide Series		89 Ac Actinium 227.028	90 Th Thorium 232.038	91 Pa Protactinium 231.039	92 U Uranium 238.029	93 Np Neptunium 237.048	94 Pu Plutonium 244.064	95 Am Americium 243.061	96 Cm Curium 247.070	97 Bk Berkelium 247.079	98 Cf Californium 251.080	99 Es Einsteinium [254]	100 Fm Fermium 257.095	101 Md Mendelevium 258.1	102 No Nobelium 259.101	103 Lr Lawrencium [262]	

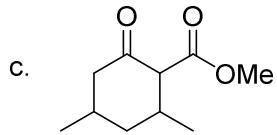
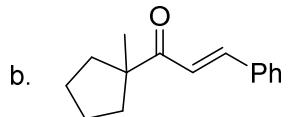
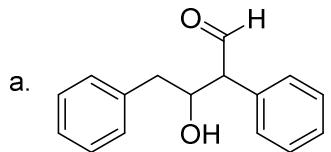
pKa Values

HI	-10	CH ₃ COOH	4.7	ArOH	10	HC≡CH	26
HBr	-8	HN ₃	4.7	RSH	10-12	H ₂	35
HCl	-6	H ₂ S	7.0	H ₂ O	15.7	NH ₃	36
H ₃ O ⁺	-1.7	NH ₄ ⁺	9.3	ROH	16-18	H ₂ C=CH ₂	45
HF	3.2	HCN	9.4	O=C-CH	9-25	CH ₄	60

1) Predict the major product of the following reactions. If no reaction occurs, then write NR. Do not show stereochemistry. If an aldol-type reaction occurs, assume it only occurs once and does not involve subsequent additions. (30 pts; 3 pts each)

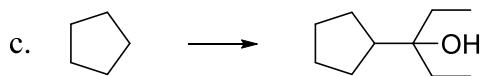
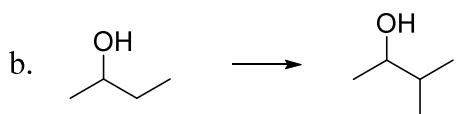
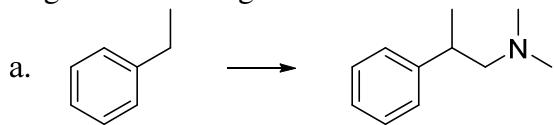


- 2) Show how you would use an aldol or Claisen reaction to make each compound. (15 pts; 5 pts each)

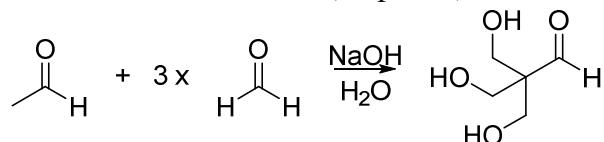


- 3) The Wieland-Miescher ketone is an important precursor in the synthesis of many steroids and other compounds including Taxol (used to treat cancer). It is synthesized from methyl vinyl ketone and 2-methyl-1,3-cyclohexanedione by Robinson annulation. Draw the mechanism for its formation and the structure of the Wieland-Miescher ketone. (15 points)

- 4) Find a way to synthesize the desired product from the given starting material plus any other reagents containing at most six carbon atoms, or triphenylphosphine. (30 pts)



- 5) Formaldehyde might look like an ideal electrophile for an aldol reaction – it cannot form an enolate and has a large partial positive charge. However, it is a little *too* reactive and is attacked by enolates too easily. If excess formaldehyde is present, the following reaction will take place. Show a mechanism for this reaction. (10 points)



- 6) Extra credit! Suggest a reasonable mechanism for this reaction. (10 points e.c.)

