Chemistry 3331-100

Organic Chemistry	/Dr. Barney	Ellison
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Name: Key (please print)

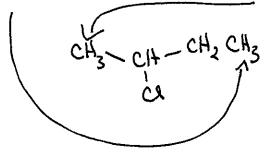
1. (10 pts) Consider the isomers of $C_4H_9C\ell$. Which species has an 1H NMR a) Contains a single peak. CH3 CACH2CH, CH3-(17) CH3-(17)

all protons one identical

b) Has several peaks including a doublet at δ 3.4.

c) Has several peaks including a triplet at δ 3.5.

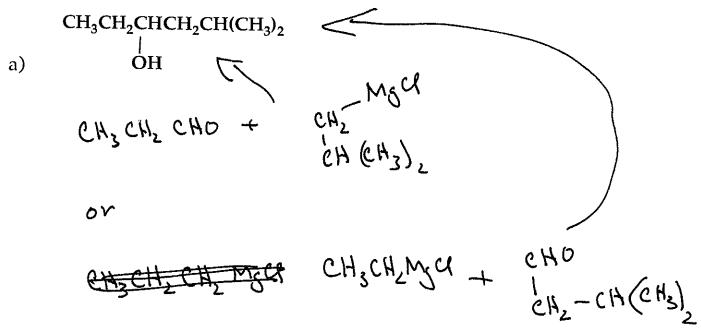
d) Has several peaks including two distinct 3-proton signals, one of them a triplet at δ 1.0 and the other a doublet at δ 1.5.



- 2. (10 pts) Using bromobenzene and any other reagents, propose a synthesis for:
- a) benzyl alcohol

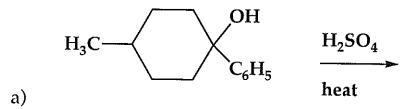
b) 1-phenylcyclooctanol

3. (10 pts) What combinations of Grignard reagent + carbonyl compounds produce:



4. (10 pts) Addition of phenylmagnesium bromide to 4-tert-butylcyclohexanone gives two isomeric tertiary alcohols as products. Both alcohols yield the same alkene when subjected to acid catalyzed dehydration. Suggest reasonable structures for these two alcohols.

5. (20 pts) Predict the products of each reaction.



b)
$$C_6H_5$$
 $a. B_2H_6$ AO_{11} C_6H_5 $b. HOOH/KOH$

$$CH_{3}COCH_{2}CH=CHCH_{2}COCH_{3}$$

$$\begin{array}{c}
1. \text{ LiA}\ell H_{4} \\
\hline
2. H_{2}O
\end{array}$$

c)

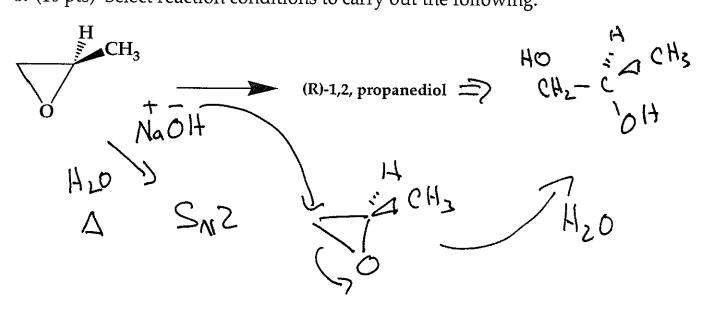
$$CH_{S} = \frac{CH - cH_{2} - CH - cH_{2}}{OH}$$

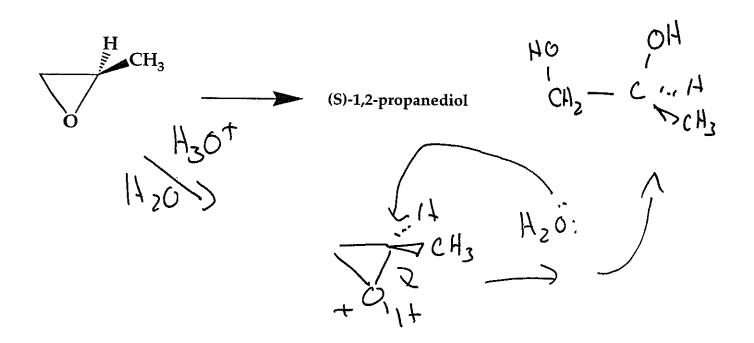
6. (10 pts) Suggest a mechanism for the pinacol rearrangement.

CH₃ CH₃
$$H_2SO_4/H_2O$$
 CH_3 C

7. (10 pts) When (R)-(+)-2-phenyl-2-butanol is allowed to stand in CH_3OH/H^+ , racemic 2-methoxy-2-phenylbutane is formed. Show a mechanism.

8. (10 pts) Select reaction conditions to carry out the following:





9. (10 pts) Suggest a mechanism for the following reaction.

$$C_{6}H_{5}CH_{2}CH_{2}CH_{2}CH_{2}CC(CH_{3})_{2} + CH_{2}=CCH_{3}$$

$$C_{6}H_{5}CH_{2}CH_{2}CH_{2}CH_{2}CH_{2}CH_{2}C-C(CH_{3})_{2}$$

$$C_{6}H_{5}CH_{2}CH_{2}CH_{2}CH_{2}CH_{2}C-C(CH_{3})_{2}$$

$$C_{6}H_{5}CH_{2}CH_{2}CH_{2}CH_{2}CH_{2}C-C(CH_{3})_{2}$$

$$C_{6}H_{5}CH_{2}CH_{2}CH_{2}CH_{2}CH_{2}C-C(CH_{3})_{2}$$

$$C_{7}H_{3}C$$

$$C_{7}H_{3}C$$

$$C_{7}H_{3}C$$

$$C_{7}H_{3}C$$

$$C_{7}H_{3}C$$

$$C_{7}H_{3}C$$

$$C_{8}H_{3}C$$

$$C_$$

$$Ph-CH_2CH_2CH=CH_2CH_2-C-CH_3$$

$$Ph-CH_2 CH_2 CH_2 - CH_2 CH_2 - CH_3$$

 $CA_2 = C-CH_3$