## Chemistry 3331-100 Organic Chemistry/Dr. Barney Ellison Thursday: Sept. $25^{rd}$ @ **7:00pm** $\rightarrow$ **9:00**/ $1^{st}$ Exam/Hale Science 230-270)

Name:	(please print)
1. (20 pts)	Outline a synthesis of each of the target compound starting with benzene
a) <i>p</i> -Cℓ-	C₀H₄-CO-CH₂CH₃

b) *m*-bromonitrobenzene

c) 2, 6-dichloro-4-nitrotoluene

d) *p*-toluene sulfonic acid

2. (10 pts) What is the product of these reactions? Show a mechanism.  $\,$ 

$$\begin{array}{c} O \\ \\ \\ \\ C \\ \\ CH_3 \\ \\ \\ CH_3COC\ell \\ \\ \\ \\ A\ell C\ell_3 \\ \\ \end{array}$$

## 3. (10 pts)

What is the mechanism of the following reaction?

$$C_6H_5$$
 $H_2SO_4$ 

4. (10 pts) What is the mechanism of the following reaction?

$$C$$
-CH  $C$ -CH

- 5. (20 pts) What are the major products of the following reactions?
- a)  $CH_2=CH-CH=CH-CH_2MgBr + D_2O \rightarrow$

$$_{\rm b)}$$
 CH<sub>2</sub>=CH-CH=CH-CH<sub>2</sub>MgBr +  $_{\rm 2}$ C CH<sub>2</sub>  $\longrightarrow$   $\longrightarrow$ 

$$\frac{Br_2}{dark} \xrightarrow{KOH} C_{10}H_8$$

## 6. (10 pts) Consider the bromination of ethylbenzene by NBS.

Why is the product  $C_6H_5$ -CH(Br)-CH $_3$ ? Why is  $C_6H_5$ -CH $_2$ -CH $_2$ Br not formed?

## 7. (10 pts)

What is the product that results when these substrates are oxidized by  $\text{MnO}_2$  in acetone solvent?

8. (10 pts) What is the mechanism for this isomerization? Why is the equilibrium to the right?