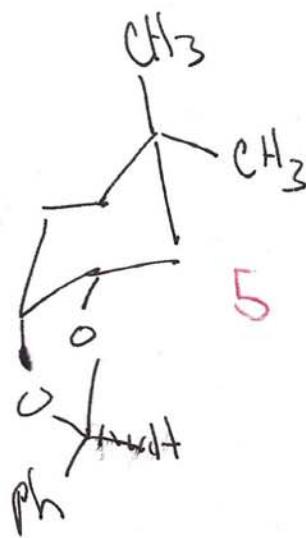
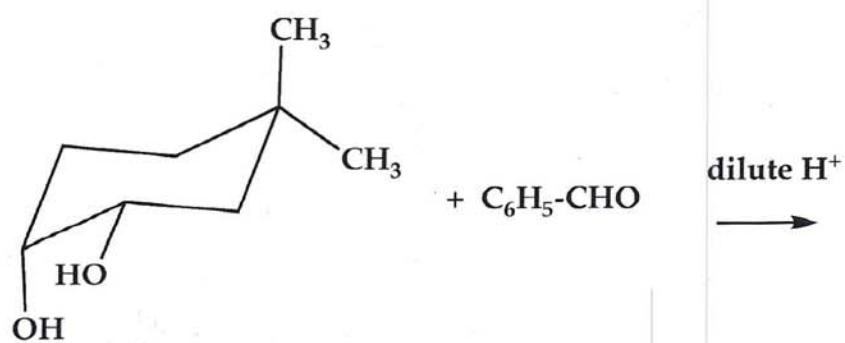


Chemistry 3331-100

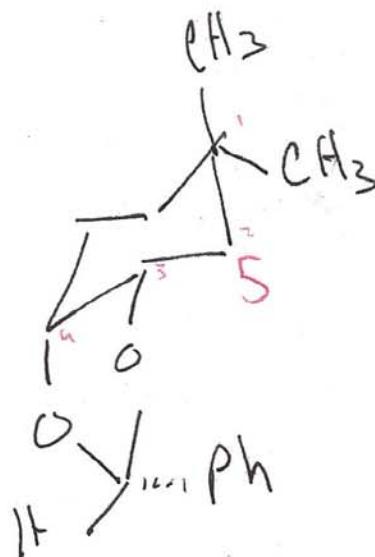
Organic Chemistry / Dr. Barney Ellison

Thursday: October 20st @ 7:00pm → 9:00 / 2nd Exam / Hale Science 270Name: Kay (please print)

1. (10 pts) What are the structures of the two separable isomers formed in the reaction:



and



HALF ACETAL -2

ESTER -5

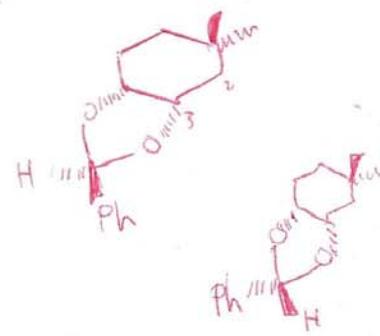
ETHER -7

OTHER -9 OR -10

STEREOCHEM. -1

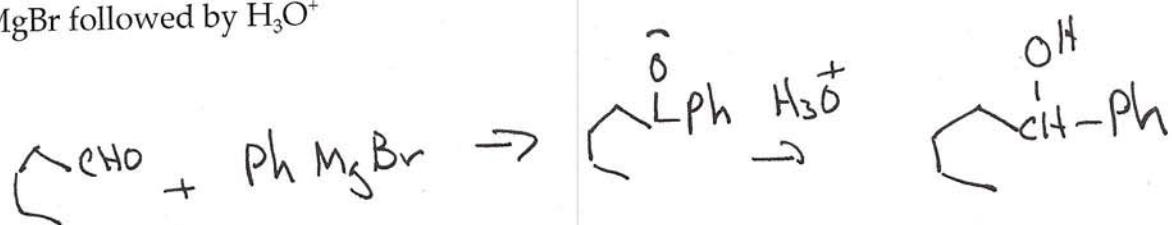
MISSING GROUP -1

CONFORMERS -3

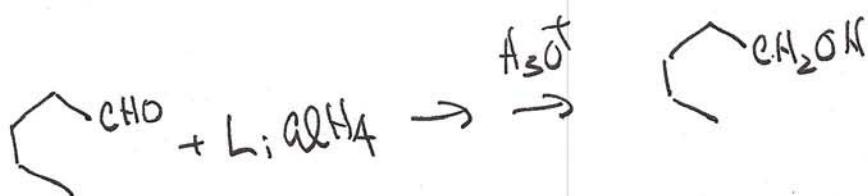


2. (15 pts) ~~Reaction~~ Give the expected products when pentanal ($\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{CHO}$) reacts with the following:

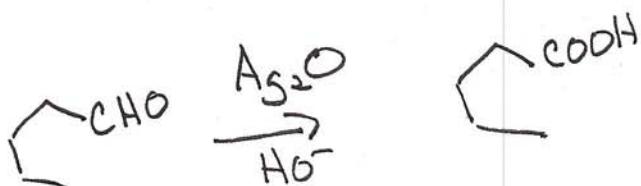
a) PhMgBr followed by H_3O^+



b) LiAlH_4 followed by H_3O^+

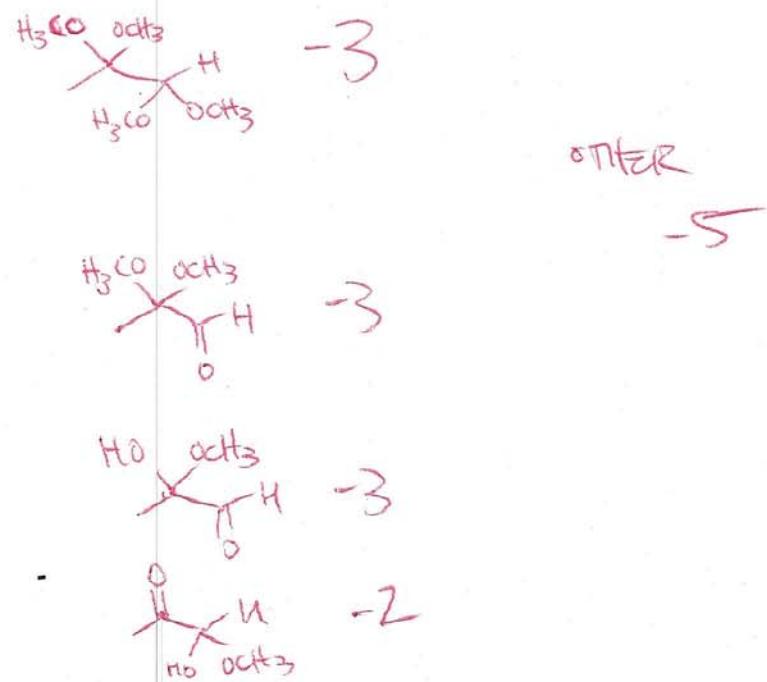
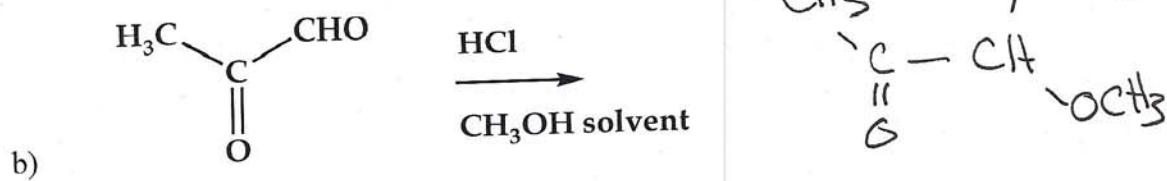
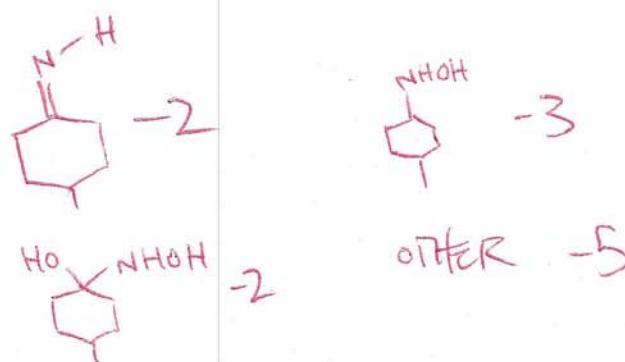
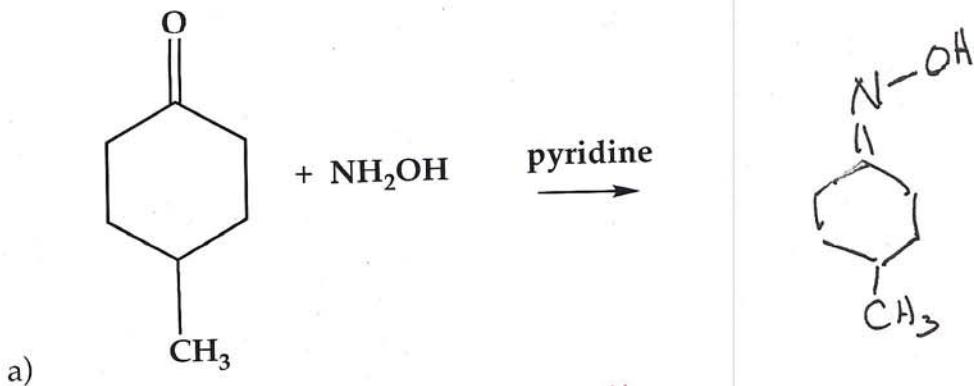


c) $\text{Ag}_2\text{O} / \text{NaOH} / \text{H}_2\text{O}$

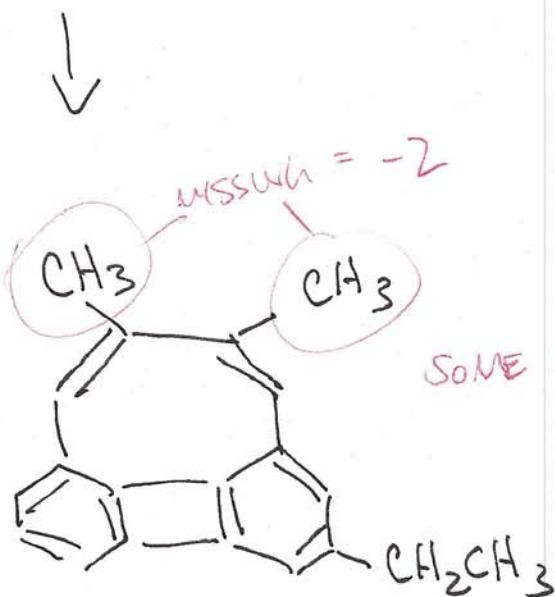
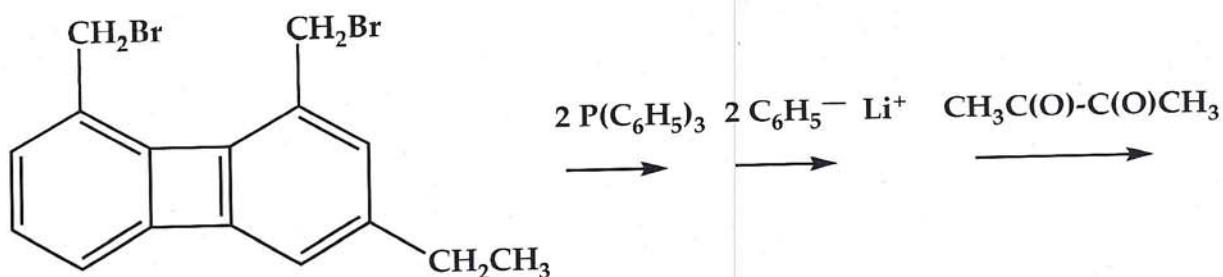


MISSING ONE ATOM -1
WRONG PRODUCT -5

3. (15 pts) Complete the reactions by giving the organic products



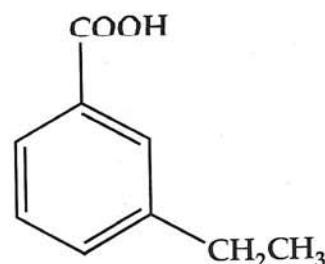
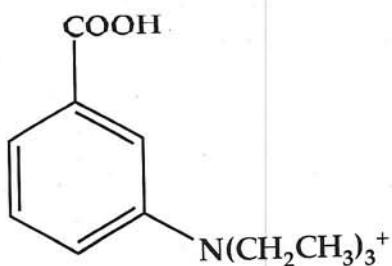
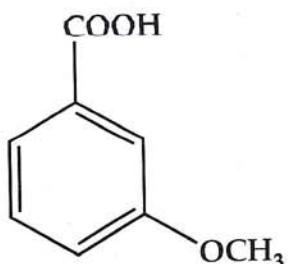
c)



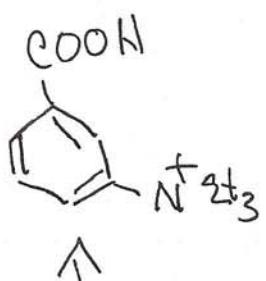
CLOSE VARIATION WITH 8-MEMBER
RNU = -4

OTHER = -5

4. (10 pts) Rank the acidity of the following compounds in order of increasing acidity.

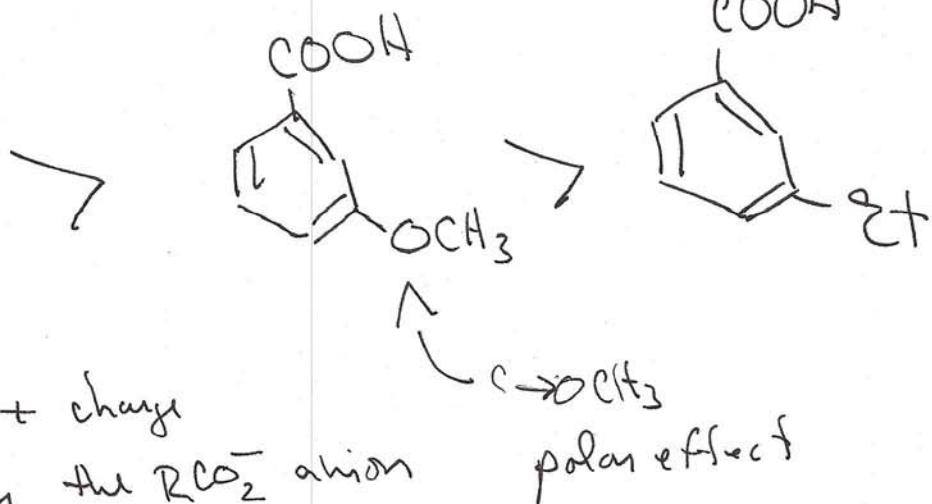


Strong acid

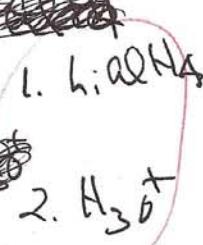
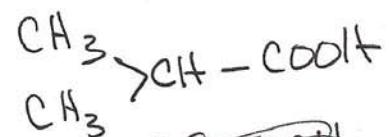
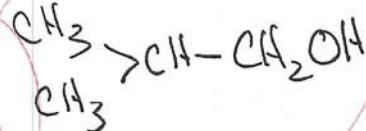
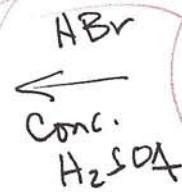
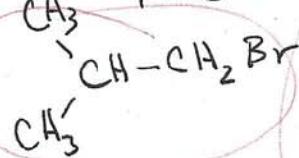
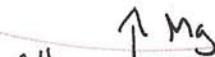
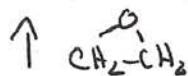
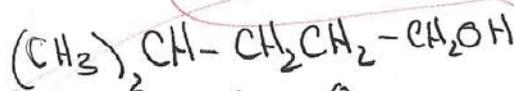
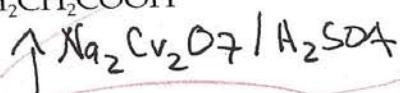


↑
because of + charge
stabilizer the RCO_2^- anion

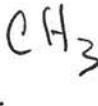
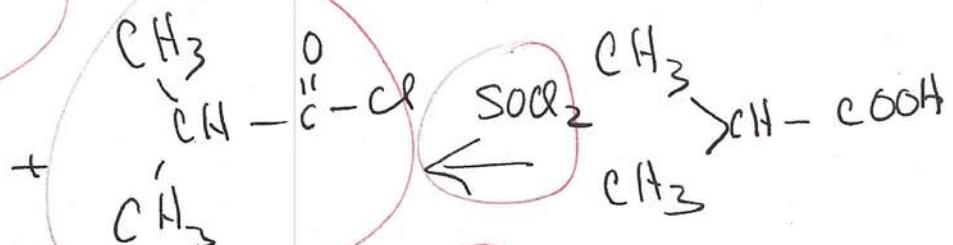
IF ANY IN WRONG ORDER = $(-3) \times 3 = -9$



5. (10 pts) Outline a synthesis of each of the following compounds from isobutyric acid (2-methylpropanoic acid)



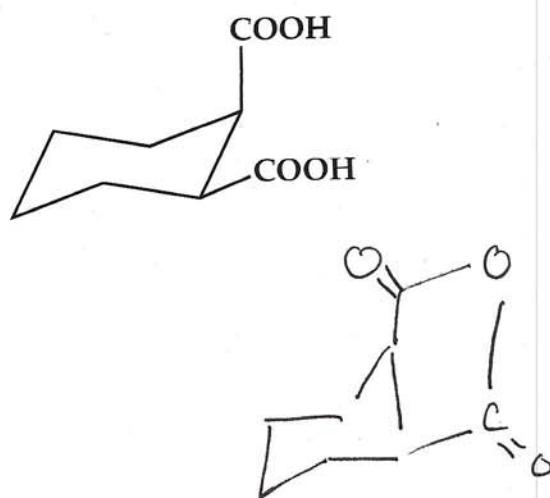
b)



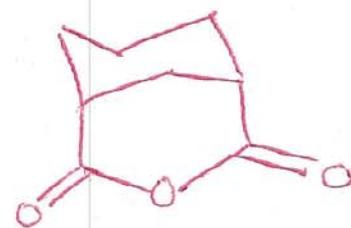
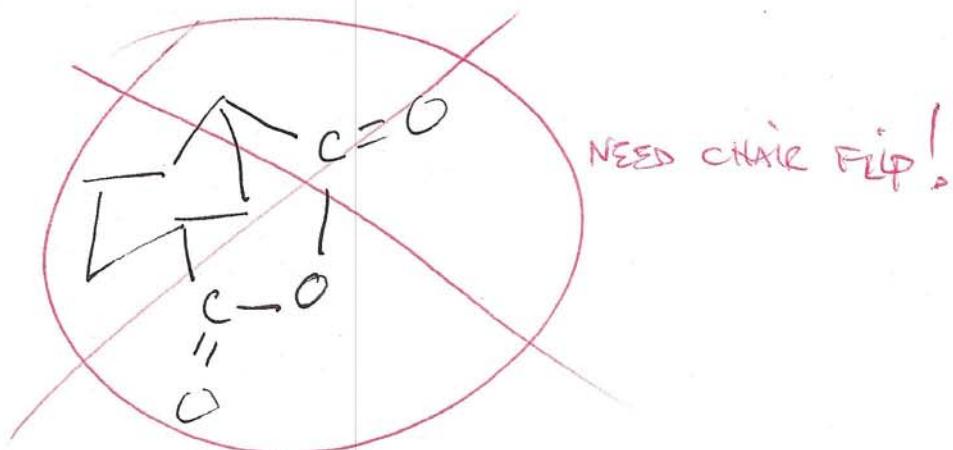
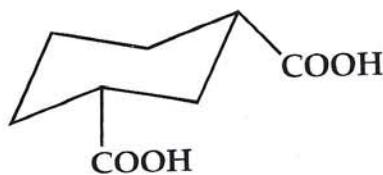
7

6. (10 pts) Draw the structure of the cyclic anhydride that forms when each is heated.

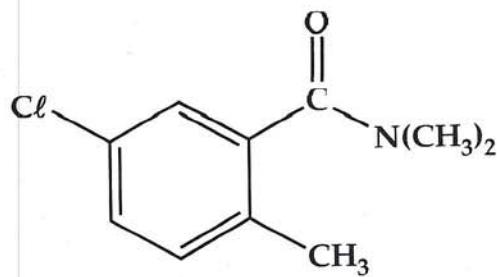
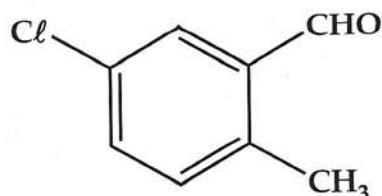
a)



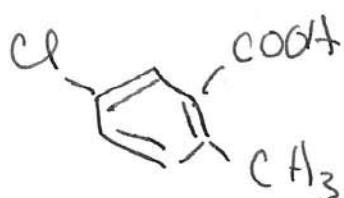
b)



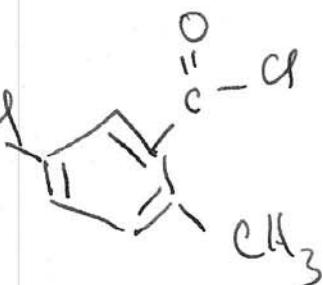
7. (10 pts) Outline a synthesis of each of the following compounds
 a)



$\downarrow \text{Na}_2\text{Cr}_2\text{O}_7$
 H_2SO_4

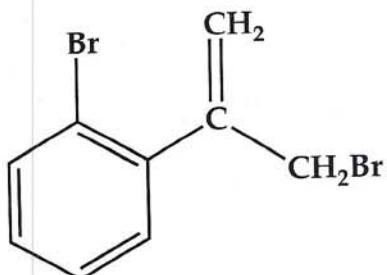
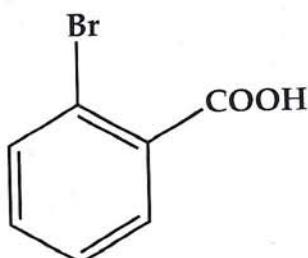


$\xrightarrow{\text{SOCl}_2}$

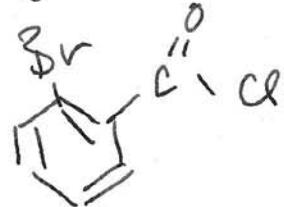


\uparrow
excess
 $\text{AN}^+ \text{---CH}_3$
 CH_3

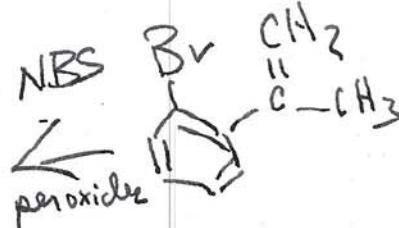
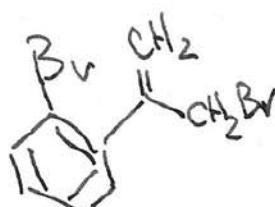
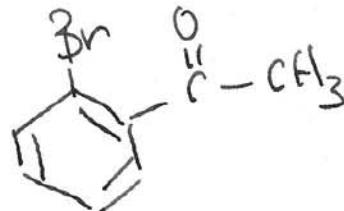
b)



$\downarrow \text{SOCl}_2$

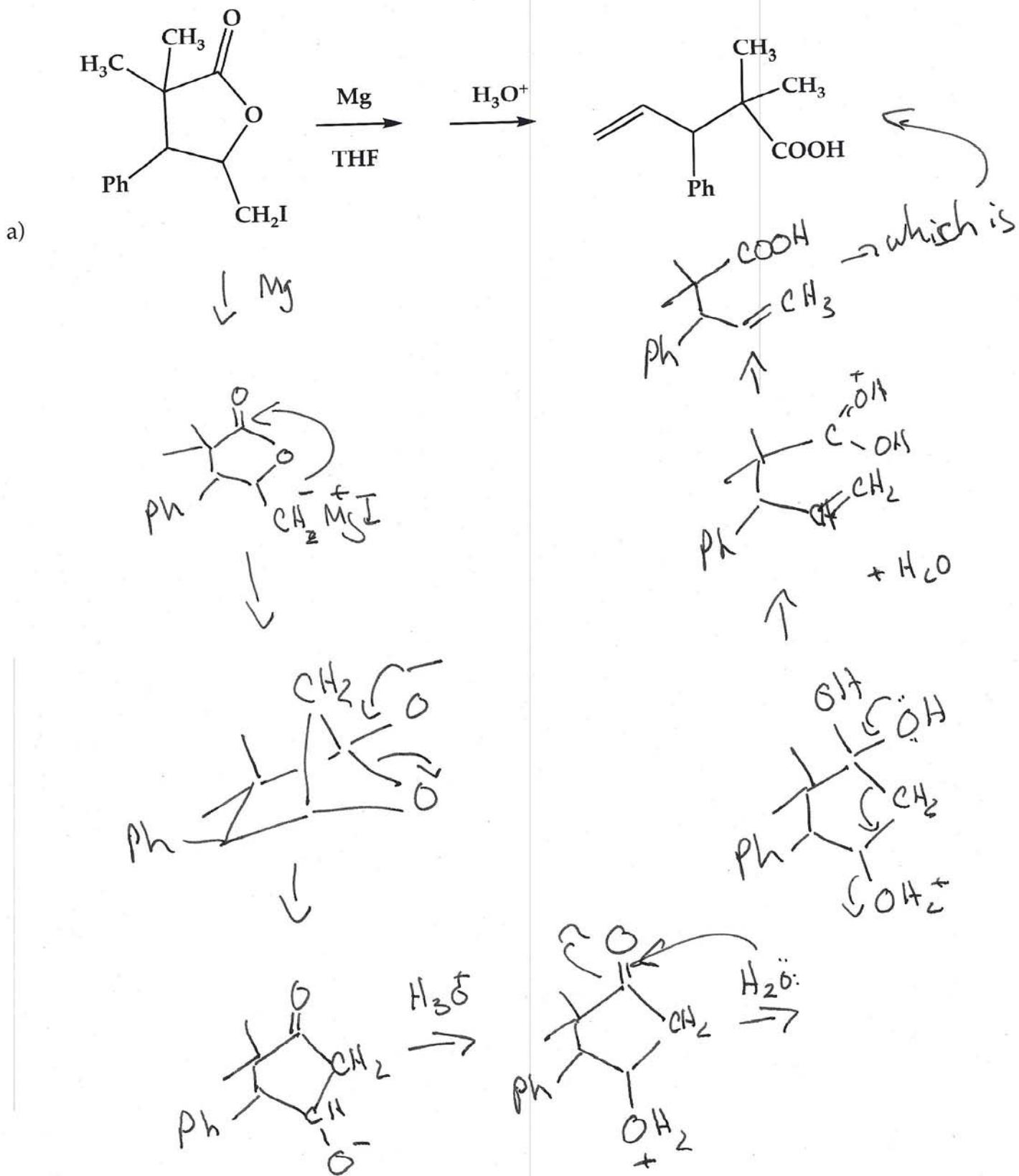


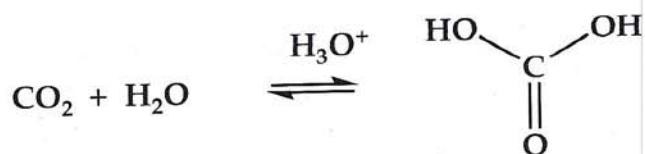
$\xrightarrow{\text{Li}^+ \text{---CH}_3^- \text{---CH}_3}$



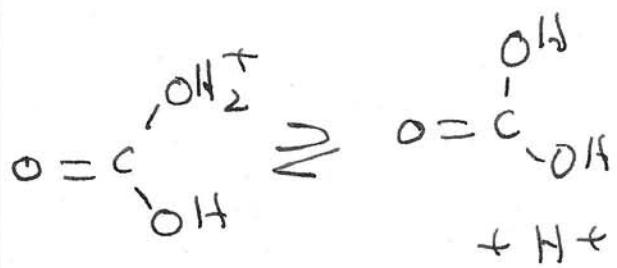
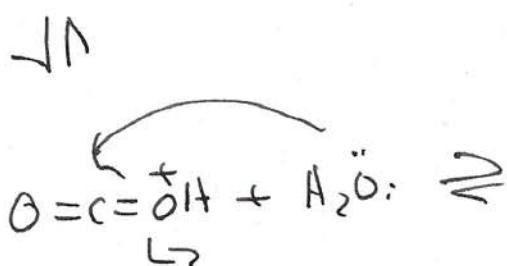
$\xrightarrow{\text{CH}_2=\text{PPh}_3}$
 w/ Tiig

8. (10 pts) Rationalize each of the reactions with a mechanism.





b)



9. (10 pts) ~~This~~ This is a Beckmann rearrangement. What is compound A? What is the mechanism of this reaction?

