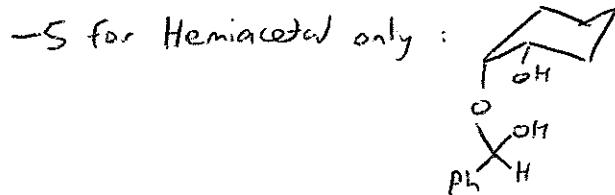
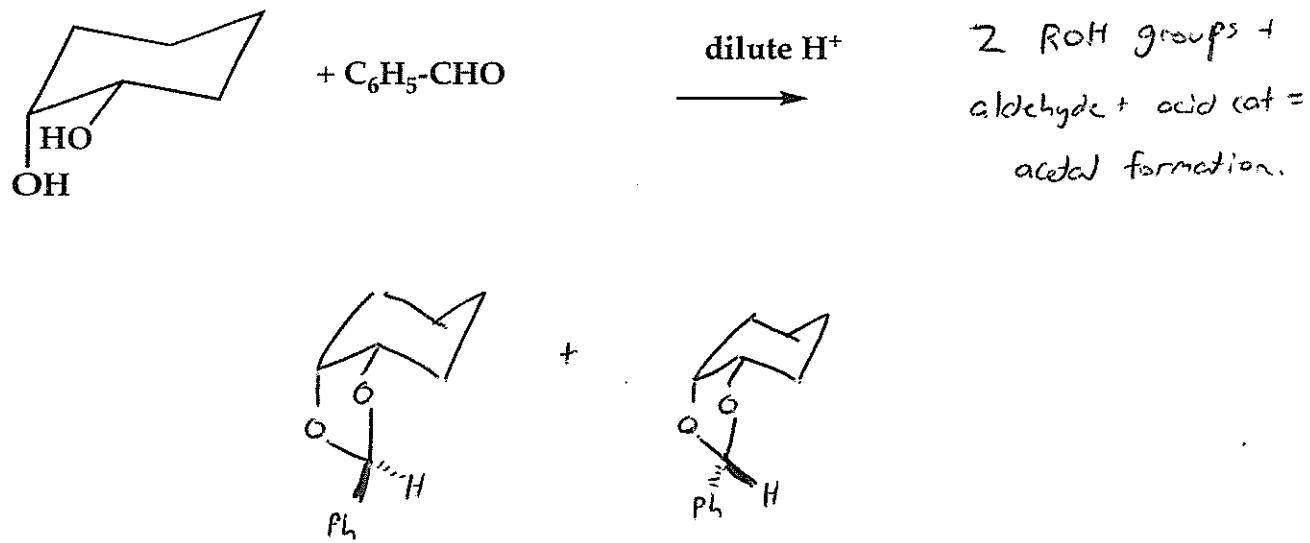


Chemistry 3331-100
Organic Chemistry/Dr. Barney Ellison
Thursday: October 21rd @ 7:00pm → 9:00 / 2nd Exam / Hale Science 230-270)

Name: Key (please print)

1. (10 pts) What are the structures of the two separable isomers formed in the reaction:



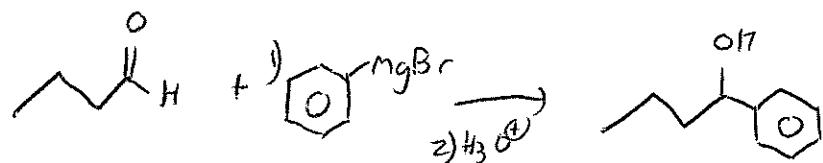
-10 for wrong product entirely.

-2.5 for no stereochem

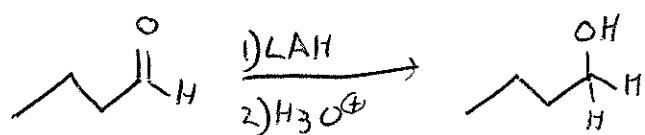
-2.5 for showing cyclohexone as benzene

2. (15 pts) Give the expected products when butyraldehyde (butanal) reacts with the following:

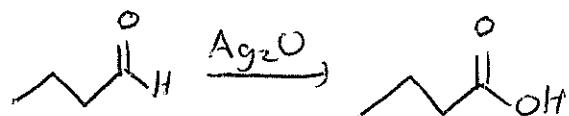
a) PhMgBr followed by H₃O⁺



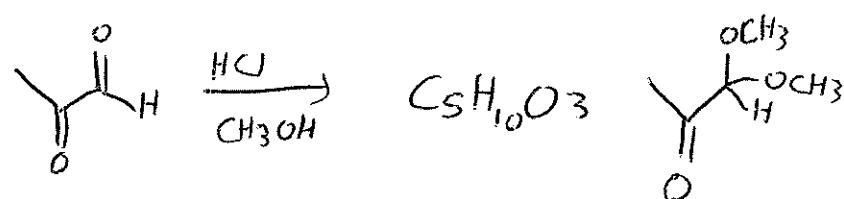
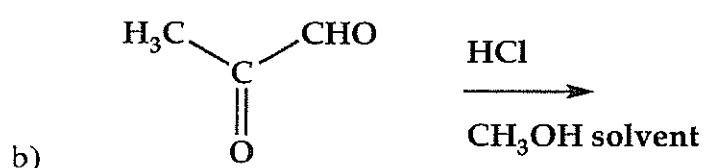
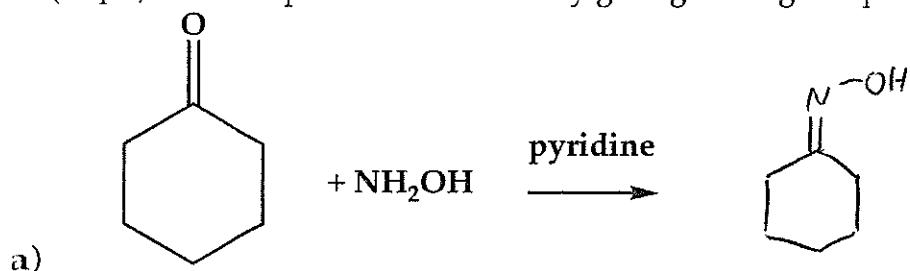
b) LiAlH₄ followed by H₃O⁺

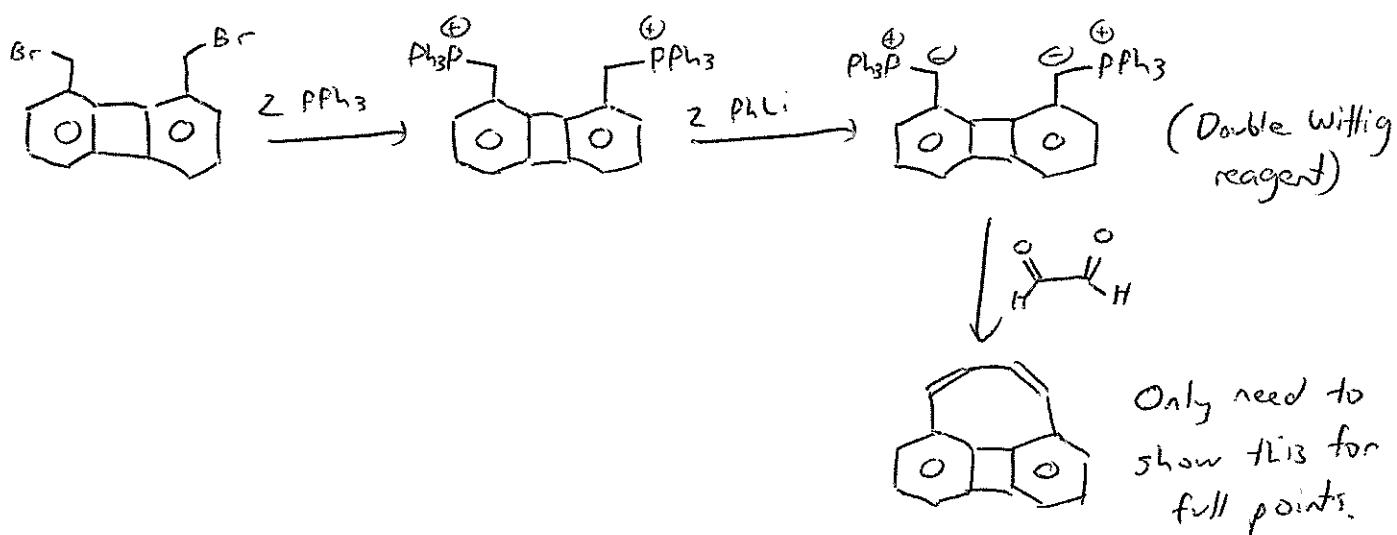
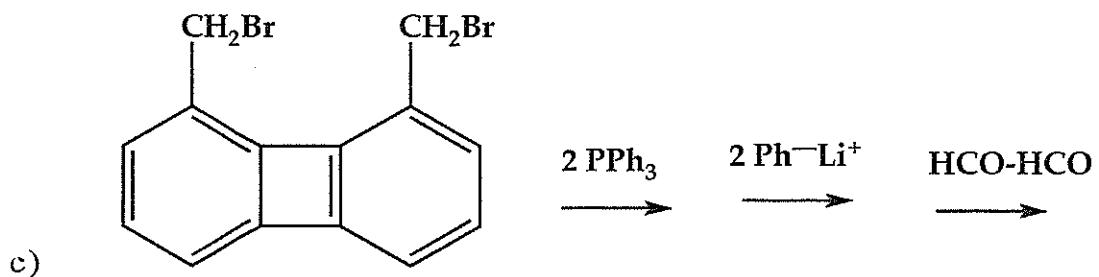


c) Ag₂O

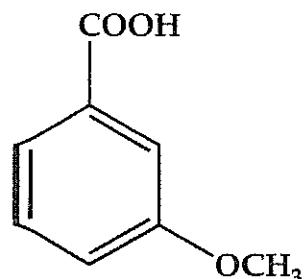


3. (15 pts) Complete the reactions by giving the organic products

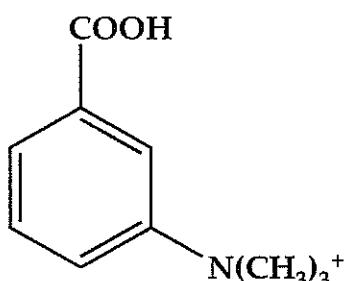




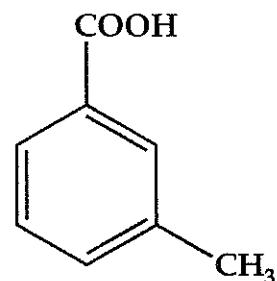
4. (10 pts) Rank the acidity of the following compounds in order of increasing acidity.



Middle



Strongest

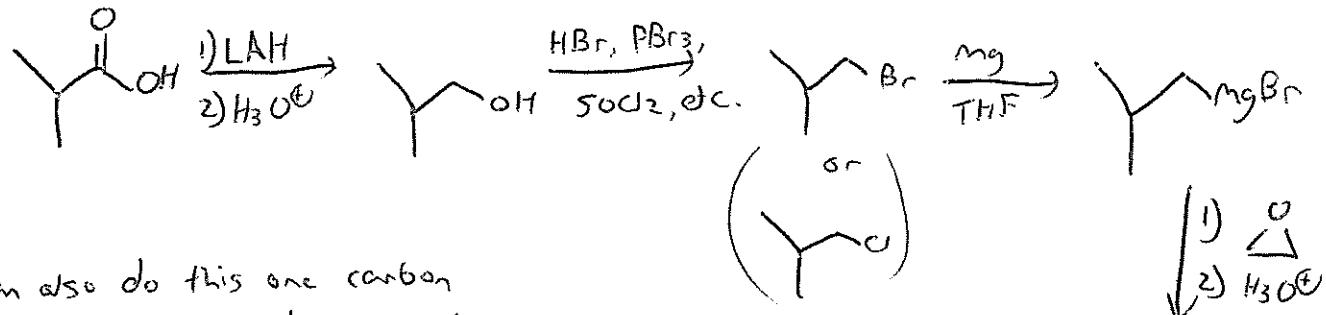


Weakest

-5 if two are wrong

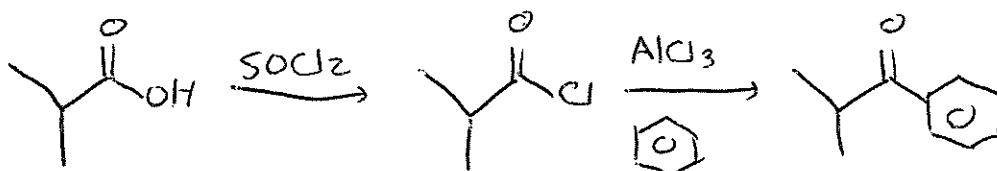
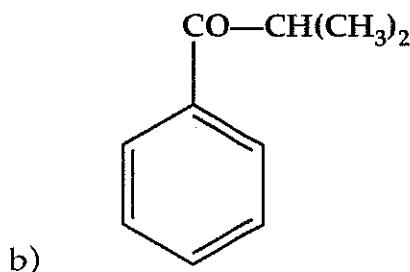
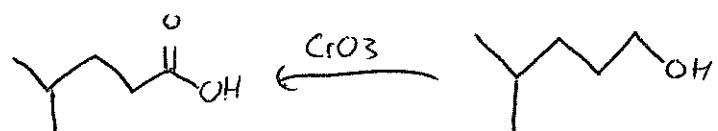
-10 if all are wrong

5. (10 pts) Outline a synthesis of each of the following compounds from isobutyric acid (2-methylpropanoic acid)



Can also do this one carbon at a time, but it's more steps.

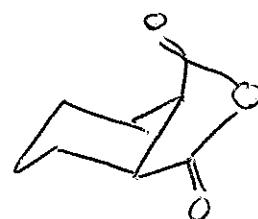
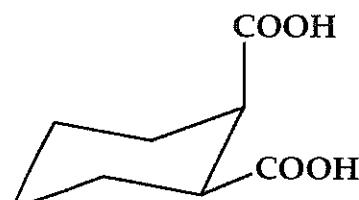
- 2 for MgBr if a carbonyl is on the molecule.
- 1 for incompatible reagents.



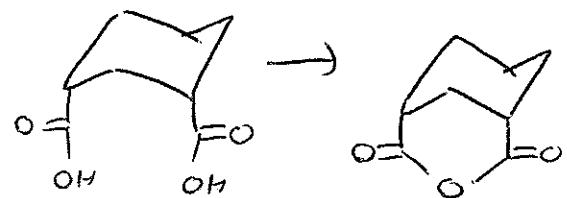
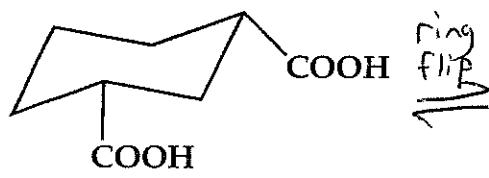
Can also do with $\text{C}_6\text{H}_5\text{MgBr}$ + $\text{CH}_3\text{CH}=\text{CH}_2$.

6. (10 pts) Draw the structure of the cyclic anhydride that forms when each is heated.

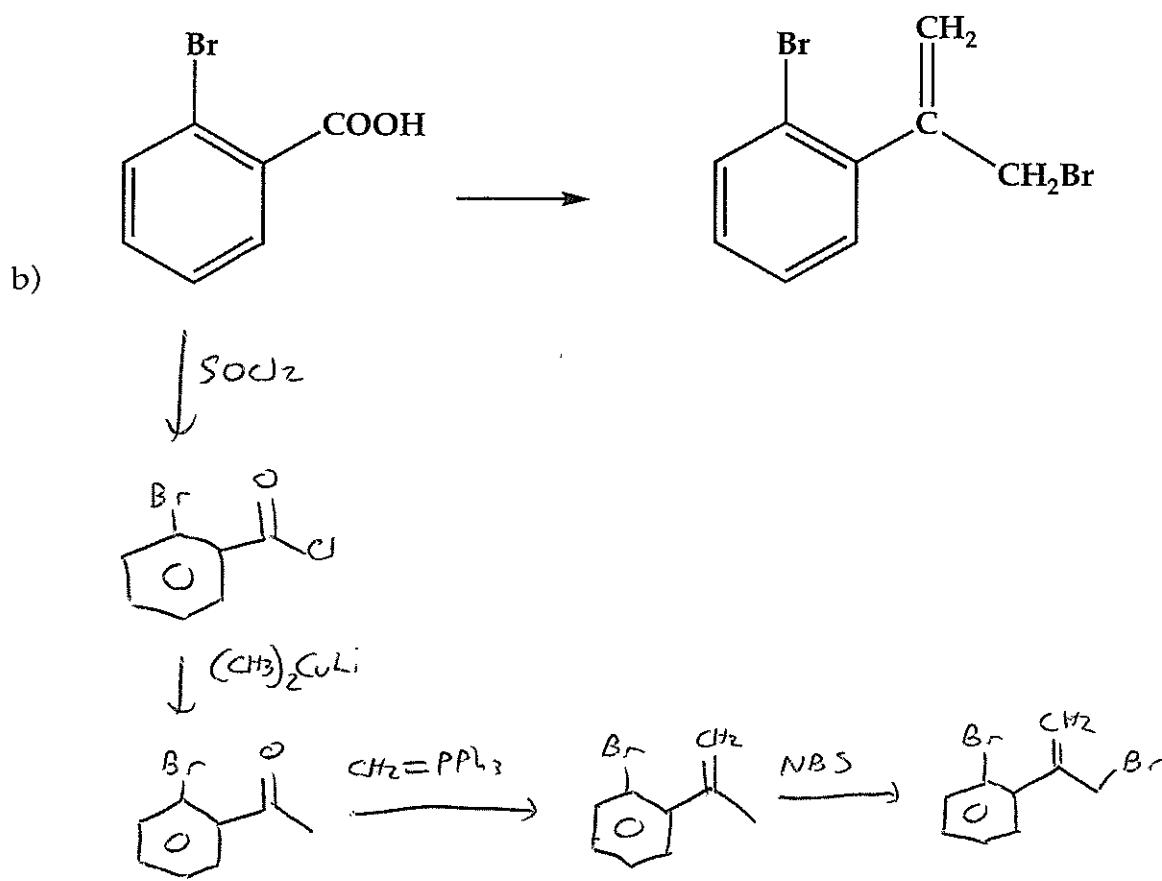
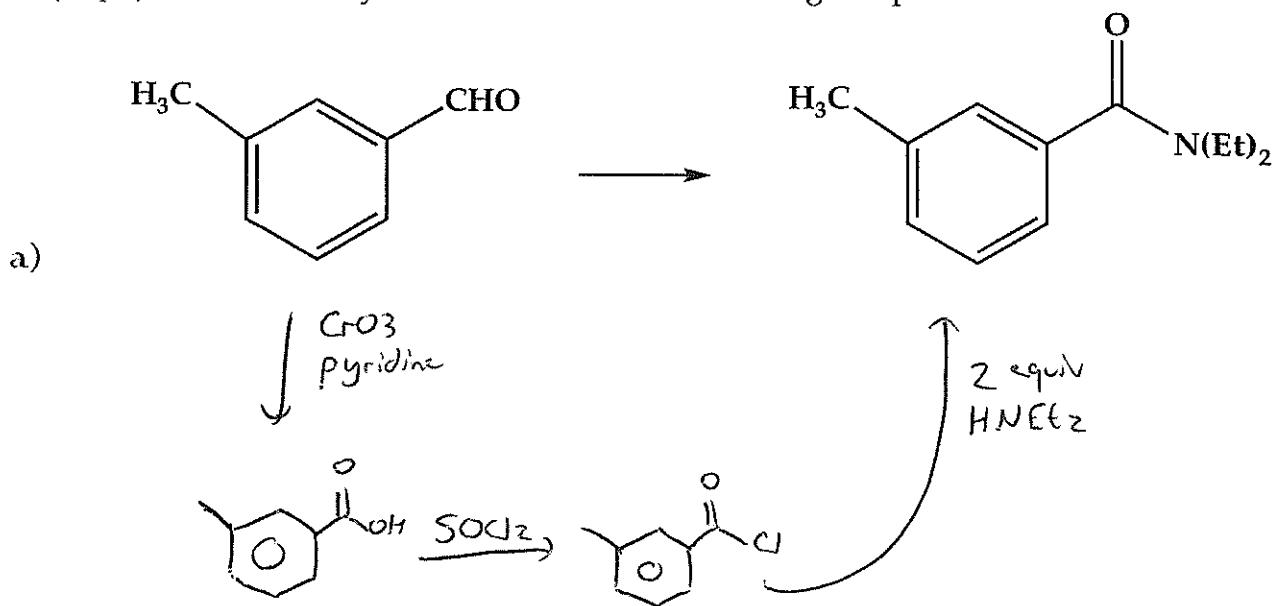
a)



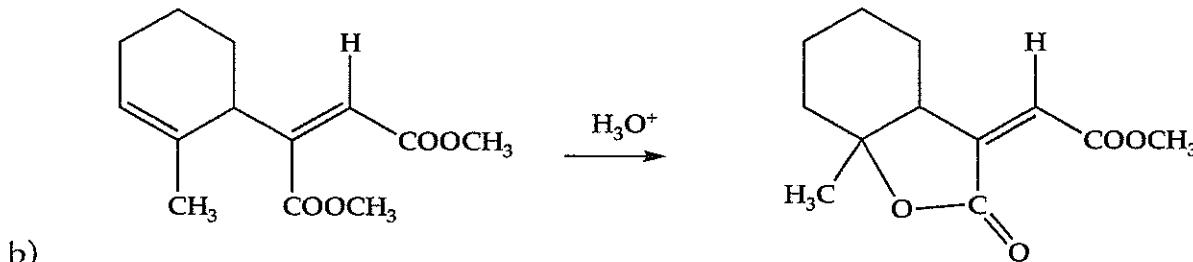
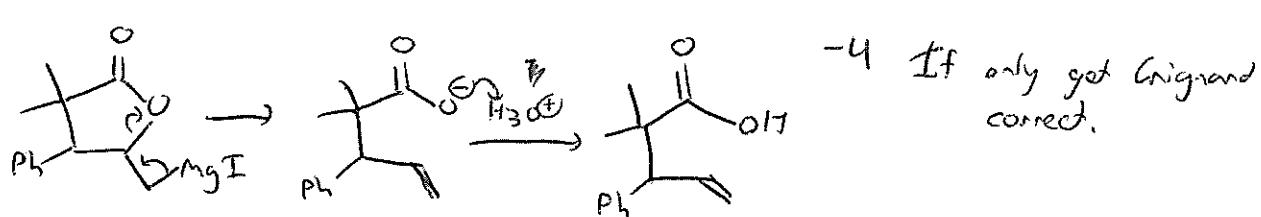
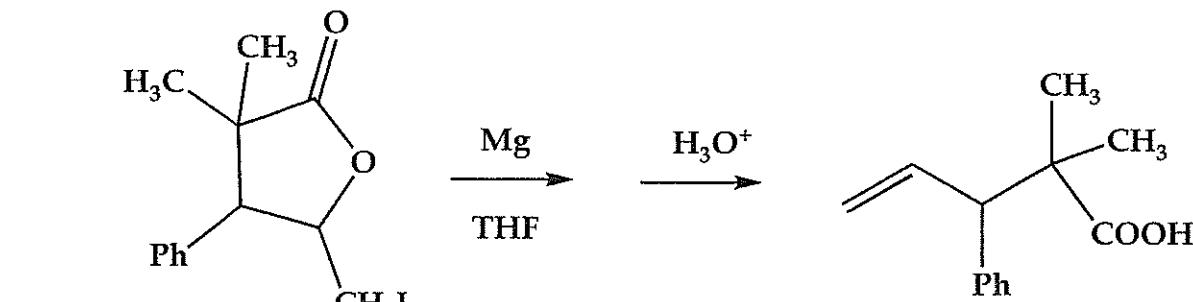
b)



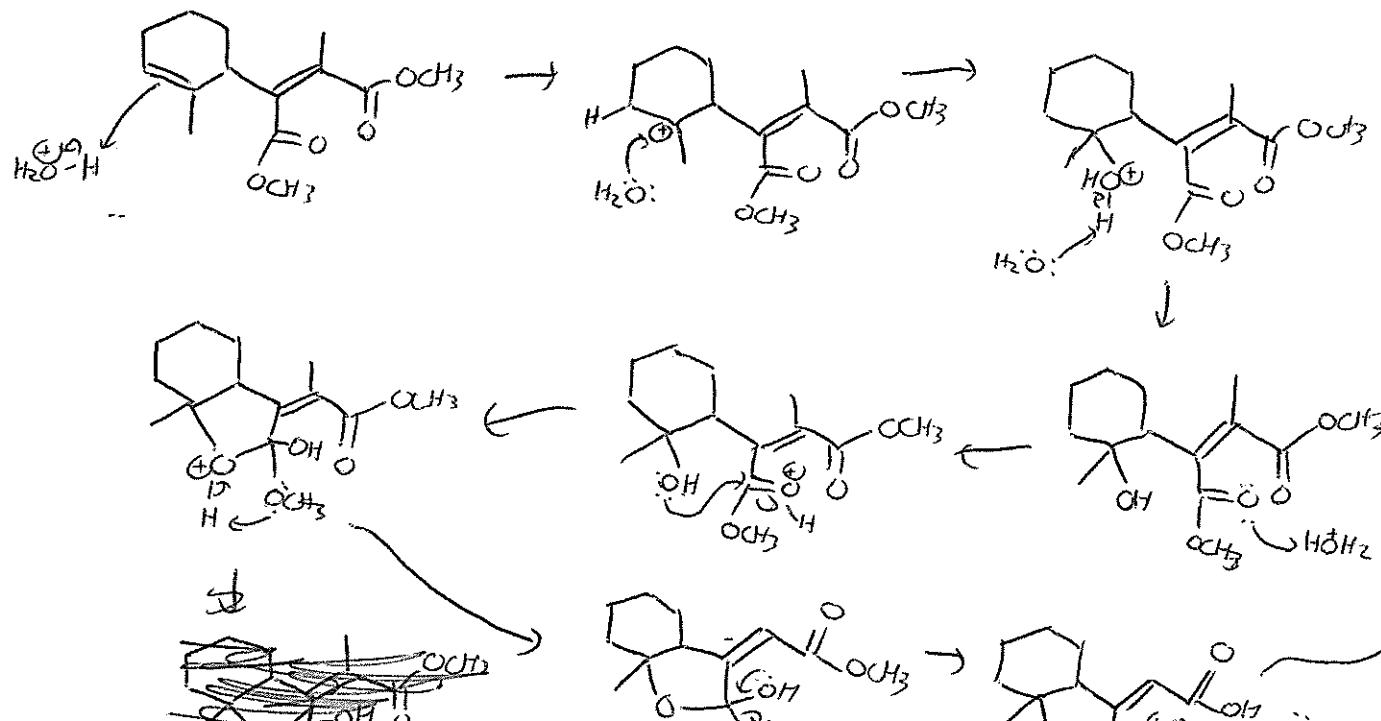
7. (10 pts) Outline a synthesis of each of the following compounds



8. (10 pts) Rationalize each of the reactions with a mechanism.



Several mechanisms are possible. Here is one:



9. (10 pts) This is a Beckmann rearrangement. What is compound A? What is the mechanism of this reaction?

