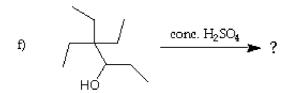
1. (30 points) Provide the missing reagents or products for the following reactions. If more than one product will result, show both the major and minor products and indicate which is the major. Carefully show the stereochemistry of each product if pertinent.

d)
$$\frac{1. \text{ BH}_3 \cdot \text{THF}}{2. \text{ H}_2\text{O}_2/\text{NaOH}} ?$$



$$_{\rm k)}$$
 $\xrightarrow{\rm Br_2}$?

$$^{\text{Br}} \xrightarrow{\text{CH}_3\text{CH}_2\text{OH}} ?$$

n)
$$\frac{\text{conc. H}_2SO_4}{}$$
 ?

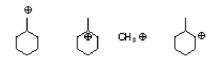
2. (20 points) Propose a synthesis of the target compounds starting with the substrate provided and any other chemical reagents. Several steps are required in each case. You do not have to show mechanisms for each of the individual steps, but do show the products formed (including any stereochemistry) from each of the reactions you use.

3. (20 points) Using the arrow formalism, draw a detailed, stepwise mechanism for the following reactions.

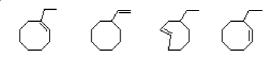
4. (30 points)

A. Rank the following $(1 \, 4, 1 = \text{greatest or most})$ according to the indicated criteria.

i. stability



ii. stability



iii. Relactivity under SN1 Conditions

iv. Stability

v. Reactivity under SN2 Conditions

B. The benzylic cation shown below is actually more stable that a tertiary carbocation even though it is formally a primary carbocation. *Very briefly* explain why this might be the case, and provide structures to bolster your reasoning.