

CHEM 3311-200 (Ellison/Richardson) 2nd Exam – March 12, 2013

Your Name

Key

Student ID No.

Recitation Day/Time

Recitation TA (circle one) Katelyn Chando,

Setareh Azamoush

Question	Score	Out of
1		15
2		8
3		10
4		10
5		12
6		10
7		10
8		15
9		10
Total		100

This is a closed-book exam. The use of notes, calculators, scratch paper, or cell phones will not be allowed during the exam. You may use models sets brought in a clear ziplock bag. Use the backs of the pages for scratch work. Please put all your final answers on the test in pen, not pencil. If your final answer is not clearly specified, you will lose points. For mechanisms, show all intermediates including correct formal charges, but do not show transition states.

Periodic Table																Period 2							
1 H																	2 He						
3 Li	4 Be																	5 B	6 C	7 N	8 O	9 F	10 Ne
11 Na	12 Mg																	13 Al	14 Si	15 P	16 S	17 Cl	18 Ar
19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr						
37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe						
55 Cs	56 Ba	57-70 *	71 Lu	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn					
87 Fr	88 Ra	89-102 **	103 Lr	104 Rf	105 Db	106 Sg	107 Bh	108 Hs	109 Mt	110 Uun	111 Uuu	112 Uub	113 Uuq	114 Uuq									

* Lanthanide series

57 La	58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb
89 Ac	90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No

** Actinide series

pKa Values

HI	-10.1	H ₂ O	15.7
HCl	-3.9	Alcohol (ROH)	16-18
H ₃ O ⁺	-1.7	HC≡CH	26
CH ₃ COOH	4.7	NH ₃	36
NH ₄ ⁺	9.3	H ₂ C=CH ₂	45
Phenol	10	CH ₄	60

Average: 67.9

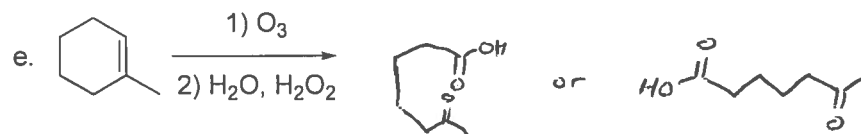
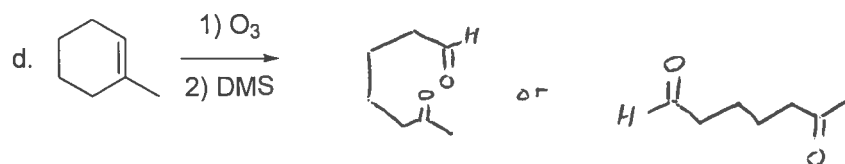
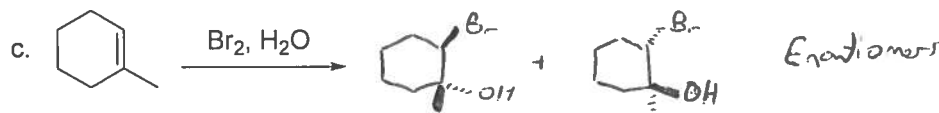
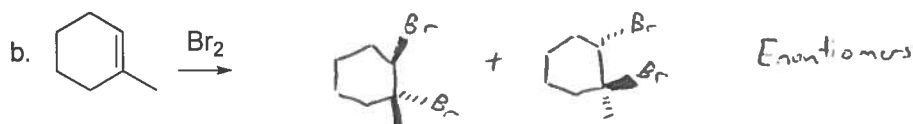
Median: 75.0

St Dev: 22.4

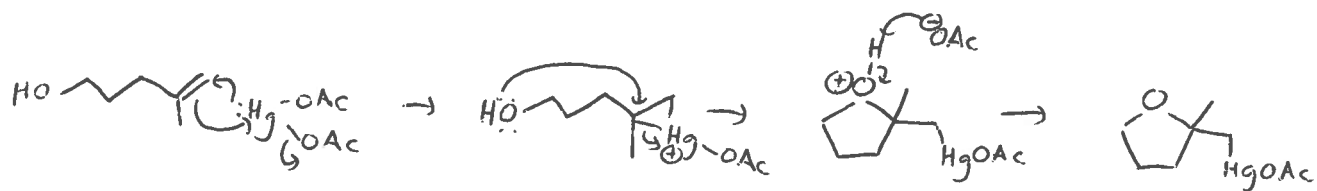
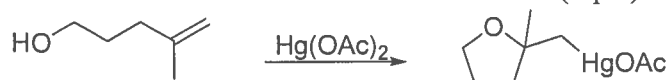
Max: 96

Min: 15

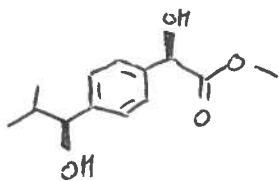
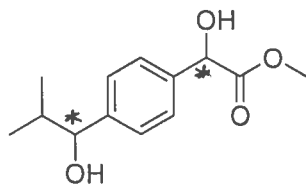
- 1) For each reaction shown below, predict the product(s). If a mixture of stereoisomers is formed, show all stereoisomers using wedges and dashes to indicate configuration, and specify whether they are enantiomers or diastereomers. (3 pts each)



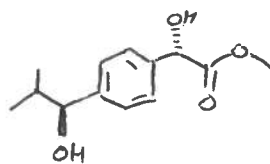
- 2) Suggest a reasonable mechanism for the reaction shown below. (8 pts)



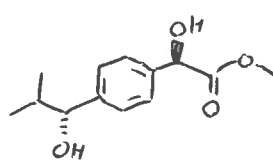
- 3) Using bold and dashed bonds, show all possible stereoisomers of the structure shown below. (4 pts)



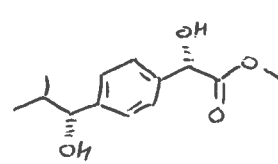
A



B



C



D

What is the stereochemical relationship (enantiomers, diastereomers, identical) between each possible pairing of these four molecules? (Hint: there are six possible pairings). (6 pts)

A & B: Diastereomers

B & C: Enantiomers

A & C: Diastereomers

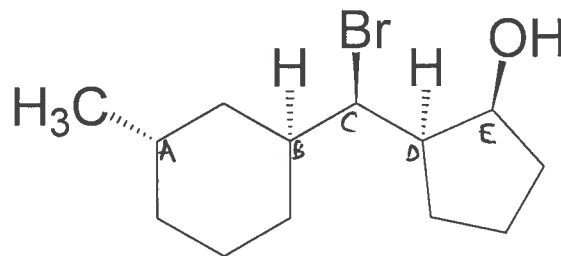
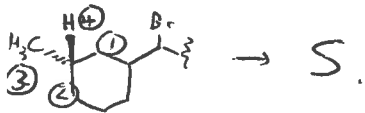
B & D: Diastereomers

A & D: Enantiomers

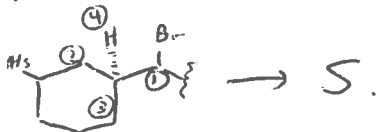
C & D: Diastereomers

- 4) In the structure shown below, label each stereocenter as R or S. (10 pts)

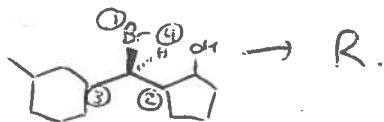
For A:



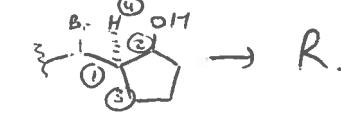
For B:



For C:



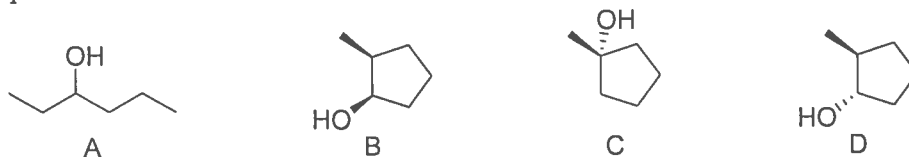
For D:



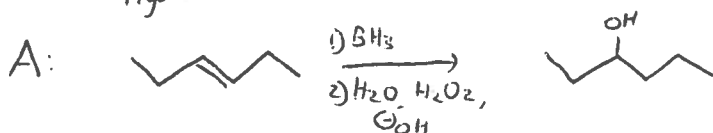
For E:



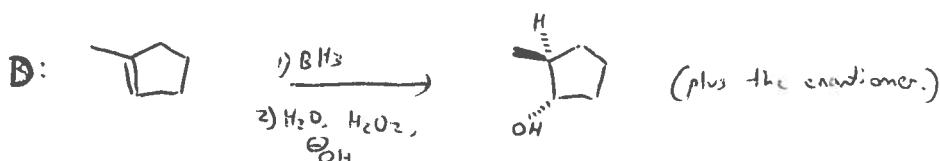
5) Four compounds are shown below.



- a. Which of these compounds can be synthesized as the major product of hydroboration-oxidation? Show the precursor alkene for each alcohol that can be prepared this way. (6 pts)
- Hydroboration-oxidation is anti-Markovnikov syn-addition.*

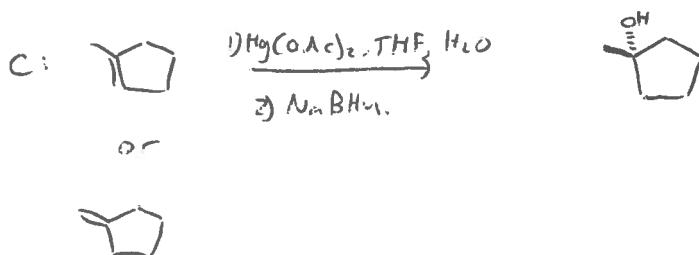
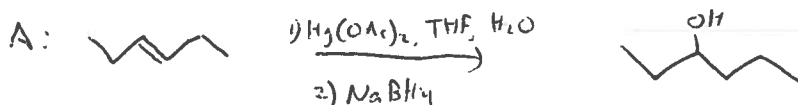


Correct letters but wrong alkene = -1
 missing a letter/extra letter = -2
 missing 2 letters/2 extra letters = -3

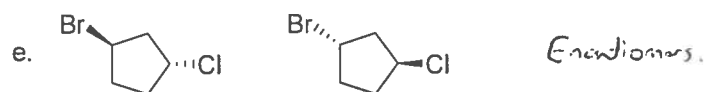
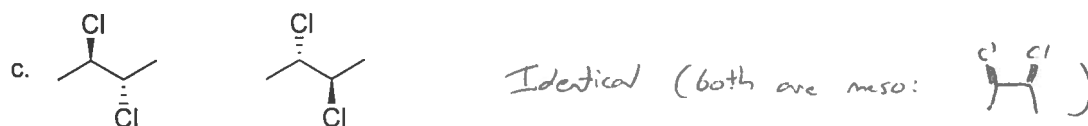
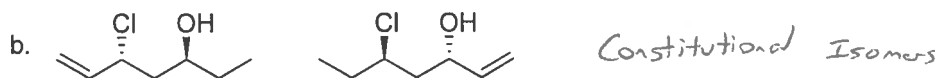
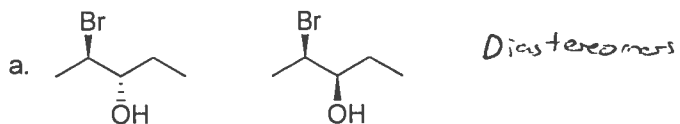


- b. Which of these compounds can be synthesized as the major product of oxymercuration-reduction? (i.e., $\text{Hg}(\text{OAc})_2$, H_2O , THF, followed by NaBH_4) Show the precursor alkene for each alcohol that can be prepared this way. (6 pts)

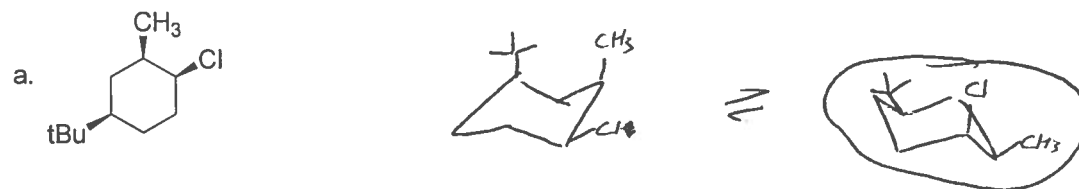
Oxymercuration-reduction is Markovnikov-style.



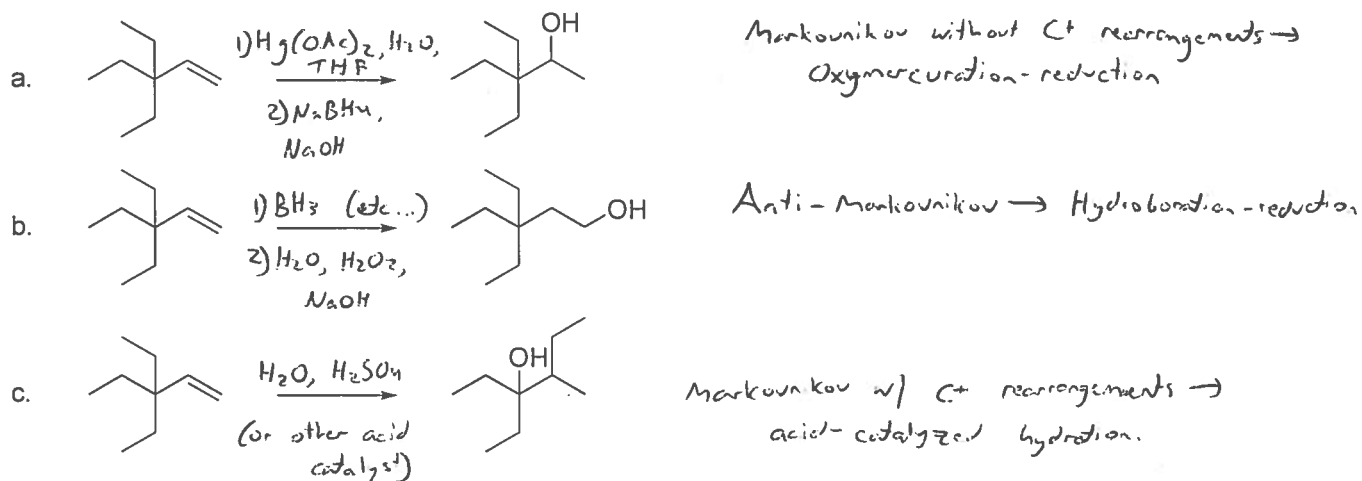
- 6) For each of the following pairs of molecules, are they identical, enantiomers, diastereomers, or constitutional isomers? (2 pts each)



- 7) For each of the following structures, show both chair conformations. (Make sure your bond angles clearly indicate whether each group is equatorial or axial.) Circle the more stable ring-flip form for each molecule. (5 pts each)

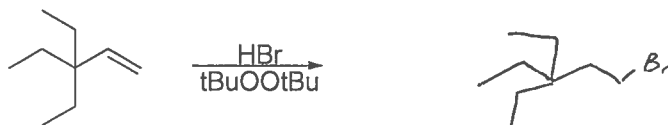


- 8) Fill in the reagents above the arrow to convert the starting material to each of the products.
(15 pts)



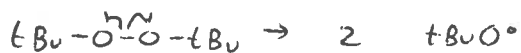
- 9) Radical HBr addition mechanism (10 pts)

- a. Predict the major product of the reaction shown below, ignoring stereochemistry. (2 pts)

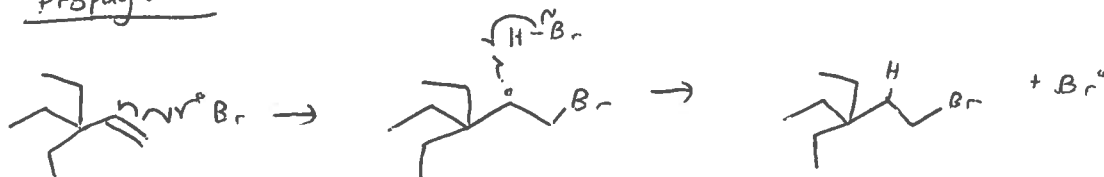


- b. Write an arrow-pushing mechanism for this reaction. Clearly label the initiation, propagation, and termination steps. Show at least two examples of termination. (8 pts)

Initiation:



Propagation:



Termination:

